

METHOD #: 272.2	Approved for NPDES and SDWA (Issued 1978)
TITLE:	Silver (AA, Furnace Technique)
ANALYTE:	CAS # Ag Silver 7440-22-4
INSTRUMENTATION:	AA
STORET No.	Total 01077 Dissolved 01075 Suspended 01076
Optimum Concentration Range:	1-25 $\mu\text{g/L}$
Detection Limit:	0.2 $\mu\text{g/L}$

1.0 Preparation of Standard Solution

- 1.1 Stock Solution: Prepare as described under "direct aspiration method".
- 1.2 Prepare dilutions of the stock solution to be used as calibration standards at the time of analysis. These solutions are also to be used for "standard additions".
- 1.3 The calibration standard should be diluted to contain 0.5% (v/v) HNO₃.

2.0 Sample Preservation

- 2.1 For sample handling and preservation, see part 4.1 of the Atomic Absorption Methods section of this manual.

3.0 Sample Preparation

- 3.1 Prepare as described under "direct aspiration method". Sample solutions for analysis should contain 0.5% (v/v) HNO₃.

4.0 Instrument Parameters (General)

- 4.1 Drying Time and Temp: 30 sec-125°C.
- 4.2 Ashing Time and Temp: 30 sec-400°C.
- 4.3 Atomizing Time and Temp: 10 sec-2700°C.
- 4.4 Purge Gas Atmosphere: Argon
- 4.5 Wavelength: 328.1 nm
- 4.6 Other operating parameters should be set as specified by the particular instrument manufacturer.

5.0 Analysis Procedure

- 5.1 For the analysis procedure and the calculation, see "Furnace Procedure" part 9.3 of the Atomic Absorption Methods section of this manual.

6.0 Notes

- 6.1 The above concentration values and instrument conditions are for a Perkin-Elmer HGA- 2100, based on the use of a 20 μL injection continuous flow purge gas and non-pyrolytic graphite. Smaller size furnace device or those employing faster rates of atomization can be operated using lower atomization temperatures for shorter time periods than the above recommended settings.
- 6.2 Background correction may be required if the sample contains high dissolved solids.
- 6.3 The use of halide acids should be avoided.
- 6.4 If adsorption to container walls or formation of AgCl is suspected, see NOTE 3 under the Direct Aspiration Method 272.1.
- 6.5 For every sample matrix analyzed, verification is necessary to determine that method of standard addition is not required (see part 5.2.1 of the Atomic Absorption Methods section of this manual).
- 6.6 For quality control requirements and optional recommendations for use in drinking water analyses, see part 10 of the Atomic Absorption Methods section of this manual.
- 6.7 If method of standard addition is required, follow the procedure given earlier in part 8.5 of the Atomic Absorption Methods section of this manual.
- 6.8 Data to be entered into STORET must be reported as $\mu\text{g/L}$.

7.0 Precision and Accuracy:

- 7.1 In a single laboratory (EMSL), Using Cincinnati Ohio tap water spiked at concentrations of 25, 50, and 75 $\mu\text{g Ag/L}$, the standard deviations were ± 0.4 , ± 0.7 , and ± 0.9 , respectively. Recoveries at these levels were 94%, 100% and 104%, respectively.